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Description

Emulsifier-free finely dispersed systems
of the water-in-oil type

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The present invention relates to emulsifier-free finely dispersed systems of the water-in-oil type, preferably as cosmetic or dermatological preparations.

Emulsions are generally taken to mean heterogeneous systems which consist of two liquids which are immiscible or have only limited miscibility with one another, which are usually referred to as phases. In an emulsion, one of the two liquids is dispersed in the form of very fine droplets in the other liquid.

If the two liquids are water and oil and if oil droplets are finely dispersed in water, then this is an oil-in-water emulsion (O/W emulsion, e.g. milk). The basic character of a O/W emulsion is defined by the water. In a water-in-oil emulsion (W/O emulsion, e.g. butter), the principle is reversed, the base character here being determined by the oil.

In order to achieve permanent dispersion of one liquid in another, emulsions in the traditional sense require the addition of an interface-active substance (emulsifier). Emulsifiers have an amphiphilic molecular structure, consisting of a polar (hydrophilic) and a non-polar (lipophilic) molecular moiety, which are spatially separate from one another. In simple emulsions, finely dispersed droplets of one phase, surrounded by an emulsifier shell, (water droplets in W/O emulsions or lipid vesicles in O/W emulsions) are present in the second phase. Emulsifiers lower the interfacial tension between the phases by positioning themselves at the interface between the two liquids. At the phase boundary, they form oil/water interfacial films, which prevent irreversible coalescence of the droplets. Emulsions are frequently stabilized using emulsifier mixtures.

Traditional emulsifiers can, depending on their hydrophilic molecular moiety, be divided into ionic (anionic, cationic and amphoteric) and nonionic:

5 the most well known example of an anionic emulsifier is soap, which is usually the term used for the water-soluble sodium salts or potassium salts of saturated or unsaturated higher fatty acids.

Important examples of cationic emulsifiers are quaternary ammonium compounds.

10 The hydrophilic molecular moiety of nonionic emulsifiers frequently consists of glycerol, polyglycerol, sorbitans, carbohydrates and polyoxyethylene glycols, and, in most cases, is linked to the lipophilic molecular moiety via ester and ether bonds. The lipophilic molecular moiety
15 usually consists of fatty alcohols, fatty acids or isofatty acids. By varying the structure and the size of the polar and nonpolar molecular moiety, the lipophilicity and hydrophilicity of the emulsifiers can be varied within wide limits.

20 A decisive factor for the stability of an emulsion is the correct choice of emulsifiers. The characteristics of all substances present in the system are to be taken into consideration. In the case of, for example, skin care emulsions, polar oil components and, for
25 example, UV filters lead to instability. As well as the emulsifiers, therefore, other stabilizers are also used which, for example, increase the viscosity of the emulsion and/or act as a protective colloid.

Emulsions are an important type of product in the
30 field of cosmetic and/or dermatological preparations.

Cosmetic preparations are essentially used for skin care. The main aim of skin care in the cosmetics sense is to strengthen or rebuild the skin's natural function as a barrier against environmental influences
35 (e.g. dirt, chemicals, microorganisms) and against the loss of endogenous substances (e.g. water, natural fats, electrolytes). If this function becomes impaired, increased resorption of toxic or allergenic substances or infection by microorganisms may result, leading to toxic

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or allergic skin reactions.

Another aim of skin care is to compensate for the loss by the skin of grease and water caused by daily washing. This is particularly important if the natural
5 regeneration ability is inadequate. Furthermore, skin care products should protect against environmental influences, in particular against sun and wind, and delay skin ageing.

Cosmetic preparations are also used as deodo-
10 rants. Such formulations are used to control body odour which is produced when fresh sweat, which is in itself odourless, is decomposed by microorganisms.

Medicinal topical compositions usually comprise one or more medicaments in an effective concentration.
15 For the sake of simplicity, in order to distinguish clearly between cosmetic and medicinal use and corresponding products, reference is made to the legal provisions in the Federal Republic of Germany (e.g. Cosmetics Regulation, Foods and Drugs Act).

The use of customary emulsifiers in cosmetic or dermatological preparations is in itself acceptable.
20 Nevertheless, emulsifiers, like ultimately any chemical substance, may in certain circumstances cause allergic reactions or reactions based on oversensitivity of the user.
25

For example, it is known that certain light dermatoses are triggered by certain emulsifiers, but also by a variety of fats and simultaneous exposure to sun-
light. Such light dermatoses are also called "Mallorca
30 acne". There has thus been no lack of attempts to reduce the amount of customary emulsifiers to a minimum, in the ideal case even to zero.

A reduction in the required amount of emulsifier can, for example, be achieved by taking advantage of the
35 fact that very finely divided solid particles have an additional stabilizing action. The solid substance accumulates at the oil/water phase boundary in the form of a layer, as a result of which coalescence of the dispersed phases is prevented. It is not the chemical

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properties of the solid particles which are of fundamental importance here, but the surface properties.

Around 1910, Pickering prepared paraffin/water emulsions which were stabilized merely by the addition of various solids, such as basic copper sulphate, basic iron sulphate or other metal sulphates. This type of emulsion is thus also referred to as a Pickering emulsion. For this type of emulsion, Pickering postulated the following conditions:

- (1) The solid particles are only suitable for stabilization if they are significantly smaller than the droplets of the inner phase and do not have a tendency to form agglomerates.
- (2) An important property of an emulsion-stabilizing solid is also its wettability. I.e. in order to stabilize an O/W emulsion, the solid has, for example, to be more readily wettable by water than by oil.

The original forms of Pickering emulsions initially surfaced, as it were, as undesired secondary effects in a variety of industrial processes, such as, for example, in secondary oil recovery, the extraction of bitumen from tar sand and other separation processes involving two immiscible liquids and fine, dispersed solid particles. These are generally W/O emulsions which are stabilized by mineral solids. Accordingly, investigation of corresponding systems, such as, for example, the oil/water/soot or oil/water/slate dust systems was initially the focus of research activity.

Basic experiments have shown that one characteristic of a Pickering emulsion is that the solid particles are arranged at the interface between the two liquid phases where they form, as it were, a mechanical barrier against the mixing of the liquid droplets.

It is a relatively new technical development to use Pickering emulsions as a base for cosmetic or dermatological preparations.

One way of achieving solids stabilization in a cosmetic or dermatological preparation is, according to

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May-Alert (*Pharmazie in unserer Zeit* [*Pharmacy in our time*], Vol. 15, 1986, No. 1, 1-7) for example, to use emulsifier mixtures which contain both anionic and cationic surfactants. Since mixing anionic and cationic surfactants produces precipitates of insoluble, electro-neutral compounds, deliberate precipitation of these neutral surfactants in the oil/water interface makes it possible to achieve additional solids stabilization in the sense of a Pickering emulsion.

EP-A-0 686 391 describes water-in-oil emulsions which are free from surface-active substances and are stabilized only by solids. Stabilization is achieved here using spherical polyalkylsilsesquioxane particles which have a diameter of from 100 nm up to 20 μ m. According to the above, these emulsions can be referred to as Pickering emulsions.

In addition to the described Pickering emulsions, the prior art describes further emulsifier-free, finely dispersed cosmetic or dermatological preparations which are generally referred to as hydrodispersions. Hydrodispersions are dispersions of a liquid, semi-solid or solid internal (discontinuous) lipid phase in an outer aqueous (continuous) phase. Hydrodispersions, like other emulsions, are metastable systems and therefore have a tendency to convert into a state of two mutually coherent discrete phases.

In the case of hydrodispersions of a liquid lipid phase in an outer aqueous phase, stability can be ensured, for example, by constructing, in the aqueous phase, a gel structure in which the lipid droplets are stably suspended. DE-A 44 25 268 describes stable finely dispersed, emulsifier-free cosmetic or dermatological preparations of the oil-in-water type, which, in addition to one oil phase and one water phase, comprise one or more thickeners from the group consisting of acrylic polymers, polysaccharides and alkyl ethers thereof, where these thickeners must not cause any lowering of the interfacial tension.

Using hydrodispersions as a basis, DE-A 43 03 983

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discloses cosmetic or dermatological light protection formulations which comprise inorganic micropigments as UV filter substances. The formulations consist of an inner lipid phase and an outer aqueous phase and are
5 essentially free from emulsifiers. The inorganic micropigments are incorporated into the lipid phase of this hydrodispersion.

The object of the present invention was to extend the prior art to include cosmetic or dermatological
10 preparations of the water-in-oil type in which it is not necessary to use any emulsifiers of a conventional type.

Surprisingly, this object is achieved by emulsifier-free cosmetic or dermatological preparations which are finely dispersed systems of the water-in-oil
15 type, comprising

1. an oil phase,
2. an aqueous phase and
3. at least one type of micronized, inorganic pigments which
20 a) have an average particle size of less than 200 nm, and whose particles
b) have both hydrophilic and lipophilic properties, i.e. have an amphiphilic character, and thus position themselves at the water/oil
25 interface, and which
c) are selected from the group consisting of metal oxides, which
d) are optionally coated on the surface

and also
30 optionally comprising further cosmetic or pharmaceutical auxiliaries, additives and/or active substances.

The amphiphilic character of the inorganic pigments according to the invention is evident, for example, from the fact that the latter are dispersible
35 both in water and in oil.

The preparations according to the invention are mixtures of oils or oil-soluble substances and water or water-soluble components, which are stabilized by adding micronized solids particles and which do not contain an

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emulsifier in the traditional sense. One way of explaining the stability of these preparations is that the pigment particles - as shown diagrammatically in Figure 1 - attach themselves to the droplets of the disperse phase and form, as it were, a mechanical barrier, which prevents coalescence of the droplets.

The preparations according to the invention are extremely satisfactory preparations in every respect, whose aqueous/fatty phase ratio can be varied within extraordinarily wide limits.

It is advantageous to choose the average particle diameter of the pigments used to be between 1 nm and 200 nm, particularly advantageously between 5 nm and 100 nm.

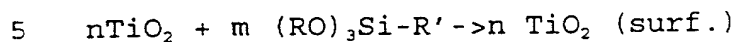
The water phase content of the formulations according to the invention is preferably chosen from the range of from 0.5 to 75% by weight, based on the total weight of the formulations.

It is essentially insignificant for the present invention in which of the potentially naturally occurring modifications the metal oxides are present.

It is advantageous for the purposes of the present invention to stabilize the Pickering emulsions using untreated, virtually pure pigment particles, in particular those which can also be used as dye in the food industry and/or as absorber of UV radiation in sunscreens. Examples of advantageous pigments are the zinc oxide pigments obtainable from Merck and those which are obtainable under the trade names zinc oxide neutral from Haarman & Reimer or NanoX from Harcros Chemical Group.

For the purposes of the present invention, Pickering emulsions are stabilized advantageously by inorganic pigments which have been surface-treated (coated) to repel water, it being the intention that the amphiphilic character is simultaneously formed or retained. This surface-treatment may involve providing the pigments with a thin hydrophobic layer by processes known per se.

One such process, which is described below using titanium dioxide as an example, consists in, for example, producing the hydrophobic surface layer according to the following reaction



n and m are arbitrary stoichiometric parameters, and R and R' are the desired organic radicals. Particularly advantageous pigments are TiO_2 pigments, for example those coated with aluminium stearate and available under the trade name MT 100 T from TAYCA.

Another advantageous coating of the inorganic pigments consists of dimethylpolysiloxane (also: dimethicone), a mixture of completely methylated, linear siloxane polymers which have been terminally blocked with trimethylsiloxy units. Zinc oxide pigments which have been coated in this way are particularly advantageous for the purposes of the present invention.

It is also advantageous to coat the inorganic pigments with a mixture of dimethylpolysiloxane, in particular dimethylpolysiloxane having an average chain length of from 200 to 350 dimethylsiloxane units, and silica gel, which is also referred to as simethicone. It is particularly advantageous for the inorganic pigments to be additionally coated with aluminium hydroxide or aluminium oxide hydrate (also: alumina, CAS No.: 1333-84-2). Particularly advantageous are titanium dioxides which have been coated with simethicone and alumina, in which case the coating can also comprise water. One example of this is the titanium dioxide obtainable under the trade name Eusolex T2000 from Merck.

For the purposes of the present invention, it is also advantageous to use a mixture of different pigment types either within a crystal, for example as a mixed iron oxide or talc (magnesium silicate), or else by combining several types of metal oxide within a preparation. Magnesium silicate, for example those obtainable under the trade name Talcum Micron from Grolmann, are particularly advantageous.

It is also advantageous, but not obligatory, to

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combine the amphiphilic inorganic micropigments according to the invention with other pigments, for example with titanium dioxide pigments which have been coated with octylsilanol, and/or with silicon dioxide particles which have been surface-treated to repel water. Silicon dioxide particles suitable for the combination are, for example, spherical polyalkylsilsesquioxane particles such as those mentioned in European laid-open specification 0 686 391. Such polyalkylsilsesquioxane particles are, for example, obtainable under the trade names Aerosil R972 and Aerosil 200V from Degussa. Suitable titanium dioxide particles are obtainable under the trade name T805, also from Degussa.

In all of the above cases it is advantageous to choose the concentration of the amphiphilic pigments according to the invention to be greater than 0.1% by weight, particularly advantageously between 0.1% and 30% by weight, based on the total weight of the preparations.

The Pickering emulsions according to the invention can be used as bases for cosmetic or dermatological formulations. These can have the customary composition and be used, for example, for the treatment and care of the skin, as lip care product, as deodorant and as make-up or make-up remover product in decorative cosmetics or as light protection preparation. For use, the cosmetic and dermatological preparations according to the invention are applied to the skin in sufficient amount in the manner customary for cosmetics.

Accordingly, for the purposes of the present invention, cosmetic or topical dermatological compositions may, depending on their structure, be used, for example, as skin-protection cream, cleansing milk, sunscreen lotion, nutrient cream, day or night cream, etc. Where appropriate, it is possible and advantageous to use the compositions according to the invention as bases for pharmaceutical formulations.

The cosmetic and dermatological preparations according to the invention may comprise cosmetic auxiliaries, as customarily used in such preparations, e.g.

preservatives, bactericides, perfumes, antifoams, dyes, pigments which have a colouring effect, thickeners, plasticizers, moisturizing and/or moisture-retaining substances, fats, oils, waxes or other customary constituents of a cosmetic or dermatological formulation, such as alcohols, polyols, polymers, foam stabilizers, electrolytes, organic solvents or silicone derivatives.

Pickering emulsions according to the invention may also comprise thickeners to improve the tactile properties of the emulsion.

In particular, the Pickering emulsions according to the invention may also comprise antioxidants. According to the invention, favourable antioxidants which can be used are any antioxidants suitable or conventional for cosmetic and/or dermatological applications.

The antioxidants are advantageously selected from the group consisting of amino acids (e.g. glycine, histidine, tyrosine, tryptophan) and their derivatives, imidazoles, (e.g. urocanic acid) and their derivatives, peptides, such as D,L-carnosine, D-carnosine, L-carnosine and their derivatives (e.g. anserine), carotenoids, carotenes (e.g. α -carotene, β -carotene, lycopene) and their derivatives, chlorogenic acid and its derivatives, lipoic acid and its derivatives (e.g. dihydrolipoic acid), aurothioglucose, propylthiouracil and other thiols (e.g. thioredoxin, glutathione, cysteine, cystine, cystamine and their glycosyl, N-acetyl, methyl, ethyl, propyl, amyl, butyl and lauryl, palmitoyl, oleyl, γ -linoleyl, cholesteryl and glyceryl esters) and their salts, dilauryl thiodipropionate, distearyl thiodipropionate, thiodipropionic acid and its derivatives (esters, ethers, peptides, lipids, nucleotides, nucleosides and salts) and sulfoximine compounds (e.g. buthionine sulfoximines, homocysteine sulfoximine, buthionine sulphones, penta-, hexa-, heptathionine sulfoximines) in very low tolerated doses (e.g. pmol to μ mol/kg), and also (metal) chelating agents (e.g. α -hydroxyfatty acids, palmitic acid, phytic acid, lactoferrin), α -hydroxy acids (e.g. citric acid, lactic

acid, malic acid), humic acid, bile acid, bile extracts, bilirubin, biliverdin, EDTA, EGTA and their derivatives, unsaturated fatty acids and their derivatives (e.g. γ -linolenic acid, linoleic acid, oleic acid), folic acid and its derivatives, ubiquinone and ubiquinol and their derivatives, vitamin C and derivatives (e.g. ascorbyl palmitate, Mg ascorbyl phosphate, ascorbyl acetate), tocopherols and derivatives (e.g. vitamin E acetate), vitamin A and derivatives (vitamin A palmitate) and coniferyl benzoate of benzoin, rutinic acid and its derivatives, α -glycosylrutin, ferulic acid, furfurylideneglucitol, carnosine, butylated hydroxytoluene, butylated hydroxyanisole, nordihydroguaiac acid, nordihydroguaiaretic acid, trihydroxybutyrophenone, uric acid and its derivatives, mannose and its derivatives, zinc and its derivatives (e.g. ZnO, ZnSO₄), selenium and its derivatives (e.g. selenomethionine), stilbenes and their derivatives (e.g. stilbene oxide, trans-stilbene oxide), and the derivatives (salts, esters, ethers, sugars, nucleotides, nucleosides, peptides and lipids) of said active substances which are suitable according to the invention.

The amount of the abovementioned antioxidants (one or more compounds) in the preparations according to the invention is preferably from 0.001 to 30% by weight, particularly preferably from 0.05 to 20% by weight, in particular 1 - 10% by weight, based on the total weight of the preparation.

If vitamin E and/or its derivatives are used as the antioxidant or antioxidants, their respective concentrations are advantageously chosen from the range of 0.001 - 10% by weight, based on the total weight of the formulation.

If vitamin A or vitamin A derivatives or carotenes or their derivatives are used as the antioxidant or antioxidants, their respective concentrations are advantageously chosen from the range of 0.001 - 10% by weight, based on the total weight of the formulation.

Cosmetic and dermatological preparations which are in the form of a sunscreen are also favourable. These preferably comprise at least one UV-A filter substance and/or at least one UV-B filter substance and/or at least one further inorganic pigment selected from the group consisting of the oxides of iron, zirconium, silicon, manganese, aluminium, cerium and mixtures thereof and also modifications in which the oxides are the active agents.

For the purposes of the present invention, it is, however, also advantageous to provide such cosmetic and dermatological preparations whose main purpose is not protection against sunlight, but which nevertheless comprise substances which protect against UV. For example, UV-A and UV-B filter substances are commonly incorporated into day cream.

The preparations according to the invention can advantageously comprise further substances which absorb UV radiation in the UV-B range, the total amount of filter substances being, for example, from 0.1% by weight to 30% by weight, preferably from 0.5 to 10% by weight, in particular from 1 to 6% by weight, based on the total weight of the preparations, in order to provide cosmetic preparations which protect the hair and/or the skin from the whole region of ultraviolet radiation.

If the emulsions according to the invention contain UV-B filter substances, the latter may be oil-soluble or water-soluble. Examples of oil-soluble UV-B filters which are advantageous according to the invention are:

- 3-benzylidenecamphor derivatives, preferably 3-(4-methylbenzylidene)camphor, 3-benzylidenecamphor;
- 4-aminobenzoic acid derivatives, preferably 2-ethylhexyl 4-(dimethylamino)benzoate, amyl 4-(dimethylamino)benzoate;
- Esters of cinnamic acid, preferably 2-ethylhexyl 4-methoxycinnamate, isopentyl 4-methoxycinnamate;
- Esters of salicylic acid, preferably 2-ethylhexyl salicylate, 4-isopropylbenzyl salicylate, homo-

menthyl salicylate;

- Derivatives of benzophenone, preferably 2-hydroxy-4-methoxybenzophenone, 2-hydroxy-4-methoxy-4'-methylbenzophenone, 2,2'-dihydroxy-4-methoxybenzophenone;
- Esters of benzalmalonic acid, preferably di(2-ethylhexyl) 4-methoxybenzalmalonate, 2,4,6-trianilino(p-carbo-2'-ethyl-1'-hexyloxy)-1,3,5-triazine.

Examples of advantageous water-soluble UV-B filters are:

- Salts of 2-phenylbenzimidazole-5-sulphonic acid, such as its sodium, potassium or its triethanolammonium salt, and also the sulphonic acid itself;
- Sulphonic acid derivatives of benzophenones, preferably 2-hydroxy-4-methoxybenzophenone-5-sulphonic acid and its salts;
- Sulphonic acid derivatives of 3-benzylidenecamphor, such as e.g. 4-(2-oxo-3-bornylidenemethyl)benzenesulphonic acid, 2-methyl-5-(2-oxo-3-bornylidenemethyl)sulphonic acid and their salts.

The list of said UV-B filters, which may be used in the Pickering emulsions according to the invention, is of course not intended to be limiting.

It can also be advantageous to use, in the Pickering emulsions according to the invention, UV-A filters which have hitherto been customarily present in cosmetic preparations. These substances are preferably derivatives of dibenzoylmethane, in particular 1-(4'-tert-butylphenyl)-3-(4'-methoxyphenyl)propane-1,3-dione and 1-phenyl-3-(4'-isopropylphenyl)propane-1,3-dione. Preparations which contain the UV-A filters are also provided by the invention. The amounts which may be used are as for the UV-B combination.

Preparations according to the invention can also be advantageously used as bases for cosmetic deodorants and antiperspirants, so that a particular embodiment of the present invention relates to Pickering emulsions as bases for cosmetic deodorants.

Cosmetic deodorants are used to control body odour which arises when fresh sweat, which is in itself odourless, is decomposed by microorganisms. Customary cosmetic deodorants are based on various modes of action.

5 In antiperspirants, astringents, mainly aluminium salts, such as aluminium hydroxychloride (aluchlorhydrate), reduce sweat production.

10 The use of antimicrobial substances in cosmetic deodorants can reduce the bacterial flora of the skin. In an ideal situation, only the microorganisms which cause the odour should be effectively reduced. The flow of sweat itself is not influenced as a result, and in ideal circumstances, only microbial decomposition of sweat is stopped temporarily.

15 The combination of astringents and antimicrobial substances in one and the same composition is also common. Antimicrobial substances are electrolytes and strong Lewis acids, and the incorporation of relatively large amounts of electrolytes and/or Lewis acids into
20 cosmetic and/or dermatological preparations is generally not without problems. Surprisingly, the Pickering emulsions according to the invention are stable even when relatively high concentrations of such active substances are used.

25 W/O Pickering emulsions according to the invention are obtainable by firstly dispersing the pigments in the fatty phase, and then combining the fatty phase with the aqueous phase. Dispersing the pigments firstly in the aqueous phase usually gives O/W Pickering
30 emulsions.

The invention thus also provides a process for the preparation of the Pickering emulsions according to the invention which is characterized in that the amphiphilic inorganic pigments are dispersed in a manner
35 known per se in the preferably liquid fatty phase, which, if desired, comprises cosmetic or pharmaceutical auxiliaries, additives and/or active ingredients, with uniform stirring and, where necessary, with heating, and during the homogenization process the aqueous phase,

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which, if desired, likewise comprises cosmetic or pharmaceutical auxiliaries, additives and/or active ingredients, is mixed with the fatty phase.

5 The following examples serve to illustrate the present invention, without limiting it. The numerical values in the examples indicate percentages by weight, based on the total weight of the respective preparations.

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EXAMPLES

	Example 1	Example 2	Example 3	Example 4
Caprylic/capric triglyceride	20	20	-	20
Mineral oil	-	-	20	
Octyldodecanol	20	15	20	20
Synthetic beeswax	2			2
Preservative	0.5	0.5	0.5	0.5
Butylmethoxydi-benzoylmethane	-	2	-	-
Octyltriazone	-	4	-	-
3-(4-methyl-benzylidene) camphor	-	4	-	-
TiO ₂ (Eusolex T2000)	4	4	6	-
ZnO (coated)	-	-	-	10
Talc	-	-	2	-
Aerosil R972	-	1	-	-
Glycerol	10	3	3	3
NaCl	1	-	-	-
Phenylbenzimidazol-sulphonic acid	-	4	-	-
NaOH	-	0.75	-	-
Water	ad 100	ad 100	ad 100	ad 100

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